

## CO<sub>2</sub> hydrogenation selectivity shift over In–Co binary oxides catalysts

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The hydrogenation of CO<sub>2</sub> into methanol by renewable H<sub>2</sub> has attracted much attention due to its mitigation of overall CO<sub>2</sub> emissions and the simultaneous production of valuable chemical products. In<sub>2</sub>O<sub>3</sub> is a promising catalyst for CO<sub>2</sub> to methanol. Introducing metal elements into In<sub>2</sub>O<sub>3</sub> (M/In<sub>2</sub>O<sub>3</sub>) is one of the main strategies to improve its catalytic performance. However, the mechanism and active sites of the M/In<sub>2</sub>O<sub>3</sub> catalytic system remain unclear and need to be further elucidated.

In the presentation, the In<sub>x</sub>–Co<sub>y</sub> binary oxides catalysts will be introduced for CO<sub>2</sub> hydrogenation to methanol. Much-improved performance and obvious product selectivity shift were observed. The optimized catalyst (In<sub>1</sub>–Co<sub>4</sub>) showed five times methanol yields than pure In<sub>2</sub>O<sub>3</sub>. And the cobalt-catalyzed CO<sub>2</sub> methanation activity was significantly suppressed, although cobalt was the majority of the metal element. To unravel the reason for this selectivity shift, detailed catalysts performance evaluation, together with several *in-situ* and *ex-situ* characterizations, were employed on cobalt and In–Co for comparative study. The results indicated CO<sub>2</sub> hydrogenation on cobalt and In–Co catalyst both followed the formate pathway, and In–Co reconstructed and generated a surface In<sub>2</sub>O<sub>3</sub>-enriched core-shell-like structure under a reductive atmosphere. The enriched In<sub>2</sub>O<sub>3</sub> at the surface significantly enhanced CO<sub>2</sub> adsorption capacity and well stabilized the intermediates of CO<sub>2</sub> hydrogenation. CO<sub>2</sub> and carbon-containing intermediates adsorbed much stronger on In–Co than cobalt led to a feasible surface C/H ratio, thus allowing the \*CH<sub>3</sub>O to desorb to produce CH<sub>3</sub>OH instead of being over-hydrogenated to CH<sub>4</sub>.

Prof. Limin Guo studied materials science and engineering at Shanghai Institute of Ceramics, Chinese Academy of Sciences. Then, as JSPS postdoctoral fellow, he spent two years (2010–2012) in Department of Chemistry, Tohoku University, Japan. From 2012–2016, he worked as postdoctoral researcher in International Institute for Carbon-neutral Energy Research, Kyushu University, Japan. Since 2016, he is full professor in School of Environmental Science and Engineering/China–European Union School for Clean and Renewable Energy, Huazhong University of Science and Technology, WPI visiting professor in International Institute for Carbon-neutral Energy Research, Kyushu University. Up to now, he has published more than 100 research/review papers.

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